

# The influence of F and F<sub>2</sub> on kinetics of electrons in BF<sub>3</sub>

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## Abstract

We present transport coefficients for electrons in mixtures of BF<sub>3</sub> with its radicals (specifically F and F<sub>2</sub>) for ratios of the electric field to the gas number density  $E/N$  from 1 to 1000 Td (1 Td = 10<sup>-21</sup> V m<sup>2</sup>). Our analysis of non-conservative collisions revealed a range of  $E/N$  where electron attachment to radicals significantly changes the electron kinetics compared to pure gaseous BF<sub>3</sub>. The results were obtained using simple solutions for Boltzmann's equation and Monte Carlo simulation.

**Keywords:** BF<sub>3</sub> • F • F<sub>2</sub> • Elendif • Monte Carlo code

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The interest in the research of electron transport phenomena in molecular gases arises from the contribution of electron-assisted processes in modeling phenomena of today's advanced technologies. For example, understanding electron transport in low temperature BF<sub>3</sub> plasmas [1] is necessary in modeling ion implantation devices.

To gain full understanding of plasma devices one needs to model them almost exactly [2,3]. Application of transport coefficients in plasma modeling is not trivial as swarm parameters are measured and calculated for pure buffer gas. At the same time, real plasmas have all the products of collisions of electrons in large numbers that may affect the energy distribution function [4,5] and plasma chemistry [6]. Thus, it is important to determine swarm data for the actual conditions that occur in plasma devices. One example of such work is our previous study of transport in CF<sub>4</sub> [7].

In this work we have calculated the electron transport coefficients for mixtures of BF<sub>3</sub>/F<sub>2</sub>/F in order to estimate when radicals begin to play a significant role in the electron kinetics. In order to optimize plasma processing for future generations of integrated circuits, the empirical development of the plasma processing tools has to be replaced by quantitative, predictive models containing all of the relevant physics and chemistry of plasmas. These results will be useful for such plasma models. In realistic plasmas, large abundances, sometimes even several percent, of fluorine radicals are present.

## NUMERICAL METHOD

Calculations were performed by using our Monte Carlo technique for electron transport that has been

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verified against basic swarm benchmarks [8,9]. Calculations were also made by a two term approximation of the Boltzmann's equation [10].

We also pay attention to non-conservative transport in presence of ionization and attachment. For example, the distinction should be made between [8,9] the so called bulk (*b*) and flux (*f*) transport properties such as drift velocity:

$$\vec{v}_b = \frac{d}{dt} \sum_{i=1}^n \vec{r}_i ; \vec{v}_f = \sum_{i=1}^n \vec{v}_i \quad (1)$$

The drift velocities obtained by Monte Carlo simulation calculated in real space (bulk) and in velocity space (flux) values which are obtained as  $\langle v \rangle$  and  $dx/dt$ , respectively. The differences are a result of the time varying number of electrons which, when differentiated, produces an additional term to the standard flux definition  $\langle v \rangle$ . The differences occur only when non-conservative processes are present.

The null collision technique has been applied to model the basic experiment where spatial profiles of absolute emission (excitation) have been measured to provide the data on excitation coefficients, contribution of fast ions and neutrals, non-equilibrium processes (such as equilibration distance), secondary electron yields at the cathode surface and back diffusion [11].

Monte Carlo simulations were mainly used to test the two term approximation to the Boltzmann's equation (TTA). The test of cross sections for low energies was performed by comparing the calculated and experimental drift velocities and characteristic energies ( $eD_T/\mu$ , where  $D_T$  is the transverse diffusion coefficient and  $\mu$  is mobility). The test at higher energies was made by using the ionization rate, while assuming that the ionization cross section is very accurately measured or calculated and therefore should not be modified further.

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**RESULTS AND DISCUSSION**

The basic set of cross section data for BF<sub>3</sub> was compiled by Biagi [12] and we have used this set in our previous work [13]. Some of the preliminary data along similar lines were already presented [14]. The electron drift velocity in pure BF<sub>3</sub> exhibits regions of negative differential conductivity (NDC) [15] similar to but smaller in magnitude than those in CH<sub>4</sub> and CF<sub>4</sub>. Results for mean energy shown in Figure 1 obtained by TTA are compared to MCS results at  $T = 0$  K. We show results for drift velocity obtained by different methods in Figure 1b. Agreement between Monte Carlo and TTA results are obtained below 20 Td. Large vibrational cross sections significantly affect region of negative differential conductivity (NDC) up to about 80 Td. Results obtained by TTA show effect of NDC that is slightly smaller than observed in experiment [16]. At higher  $E/N$  excel-

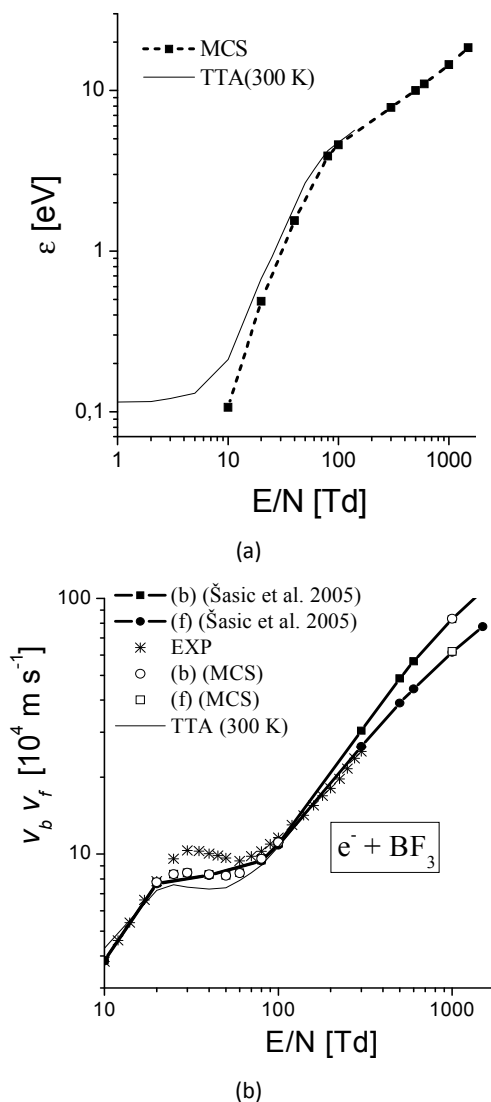


Figure 1. a) Mean energy and b) drift velocity as a function of  $E/N$  for BF<sub>3</sub>. The data for BF<sub>3</sub> from ref. 15 is denoted as EXP.

lent agreement exist between results obtained by two Monte Carlo techniques and TTA.

The electron mean energy and electron drift velocity in the mixture BF<sub>3</sub>/F<sub>2</sub> as a function of  $E/N$  are shown in Figure 2. Adding less than 0.01, 0.1, 1 and 10% of other constituents does not change significantly the drift velocities or mean energies.

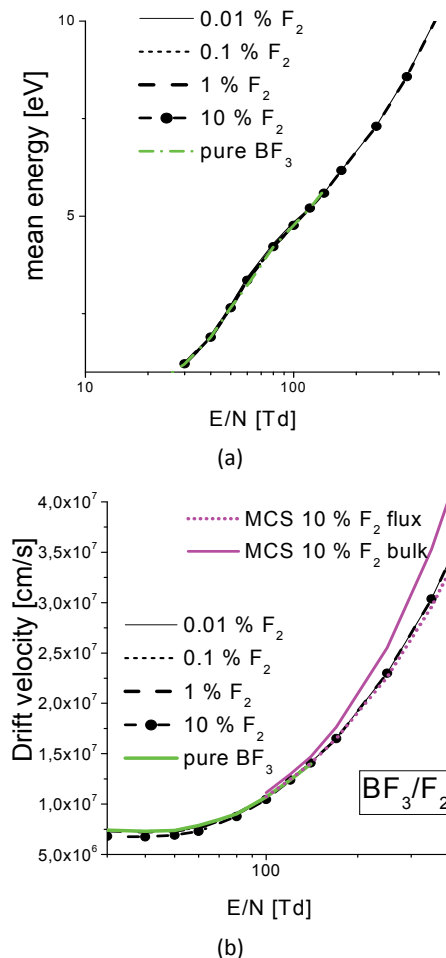


Figure 2. a) Mean energy and b) drift velocity as a function  $E/N$  for mixtures BF<sub>3</sub> and 0.01%, 0.1%, 1% and 10% F<sub>2</sub>. Results for the pure BF<sub>3</sub> are also shown.

In Figure 3 we show total attachment coefficients and effective ionization coefficient for the mixture BF<sub>3</sub>/F<sub>2</sub> as a function of  $E/N$  accounting for total electron production in the mixture. Ionization of BF<sub>3</sub> affects electron kinetics from about 100 Td. The effect of radicals on ionization is small at higher  $E/N$ . It is important to note that due to a large threshold of attachment in BF<sub>3</sub> addition of a radical with small threshold for attachment will lead to considerable changes, turning the otherwise electropositive plasma into electronegative. Thus, one needs to test the cross sections for BF<sub>x</sub> radicals in the same way as it was done for CF<sub>4</sub> [7].

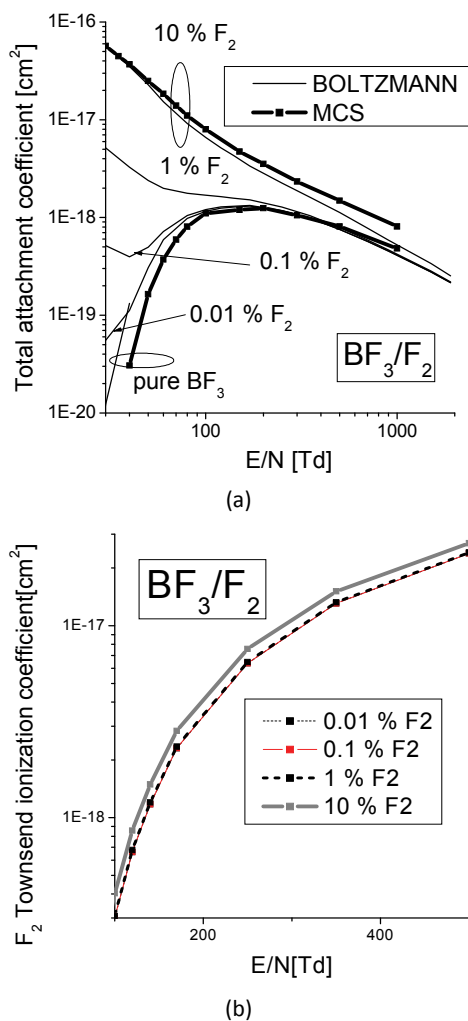


Figure 3. a) Total attachment coefficients and b) ionization coefficients in BF<sub>3</sub>/F<sub>2</sub> mixtures (0.01, 0.1, 1 and 10% of F<sub>2</sub>). Total attachment results for the pure BF<sub>3</sub> are also shown.

**CONCLUSION**

We have presented electron transport coefficients in mixture of BF<sub>3</sub>, F and F<sub>2</sub>. Effect of F<sub>2</sub> radicals on electron kinetics is relatively small for all abundances up to 10%. Only if a significant fraction of F<sub>2</sub> is added, one may expect changes in ionization and attachment rate although effective ionization rate may not be changed.

As discussed above, the effect may be important in understanding possible discrepancies between experiments and models that do not include the cross sections for electron collisions with radicals, as well as for understanding electronegative plasmas and reducing charging damage by using double layers to accelerate negative ions into charged high aspect ratio structures in dielectrics.

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**IZVOD****UTICAJ F I F<sub>2</sub> NA KINETIKU ELEKTRONA U BF<sub>3</sub>**

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U ovom radu prikazani su elektronski transportni koeficijenti za smešu F i F<sub>2</sub> sa BF<sub>3</sub> u funkciji redukovano električnog polja od 1 Td do 1000 Td (1 Td = 10<sup>-21</sup> V m<sup>2</sup>). Analiza nekonzervativnih sudara je izvršena za promene koeficijenta elektronskog zahvata, sa i bez prisustva radikala i kada je prisutan samo čist gas BF<sub>3</sub>. Rezultati su dobijeni korišćenjem jednostavnih rešenja Bolcmanove jednačine i Monte Carlo simulacija.

*Ključne reči:* BF<sub>3</sub> • F • F<sub>2</sub> • Elendif • Monte Carlo kod